



## Synthesis and biological evaluation of a series of tangeretin-derived chalcones

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### ABSTRACT

A series of chalcones polyoxygenated on the ring A (with pentamethoxy or 2'-hydroxy-3',4',5',6'-tetramethoxy substitution patterns) was synthesized from tangeretin, a natural *Citrus* flavonoid. These chalcones were evaluated for their antiproliferative, activation of apoptosis, inhibition of tubulin assembly and antileishmanial activities. Comparison with the reference analogous 3',4',5'-trimethoxylated chalcones showed that such peroxygenated substitution patterns on the ring A were less beneficial to these activities.

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Chalcones are natural or synthetic compounds bearing the 1,3-diphenylprop-2-en-1-one **1** framework, that have displayed a wide pharmacological spectrum including, among others, cytotoxic, antitumour, antiviral, and antiprotozoal activities.<sup>1</sup> Many studies of chalcones related to cancer have demonstrated the positive influence of a polymethoxylated ring A on cytotoxicity, though the optimal substitution pattern remains to be defined.<sup>2,3</sup> For instance, a 3',4',5'-trimethoxyphenyl ring A is present in the strongly cytotoxic chalcones **2**, **3**, and **5**, that interfere with the mitotic phase of the cell cycle.<sup>4–7</sup> The biological profile of **2**, **3**, and **5** can be easily related to the structural analogy of these chalcones with combretastatin A4 **6** and its amino analog **7**, two powerful inhibitors of tubulin assembly now under clinical investigation as their respective prodrugs **8** and **9**.<sup>8</sup>

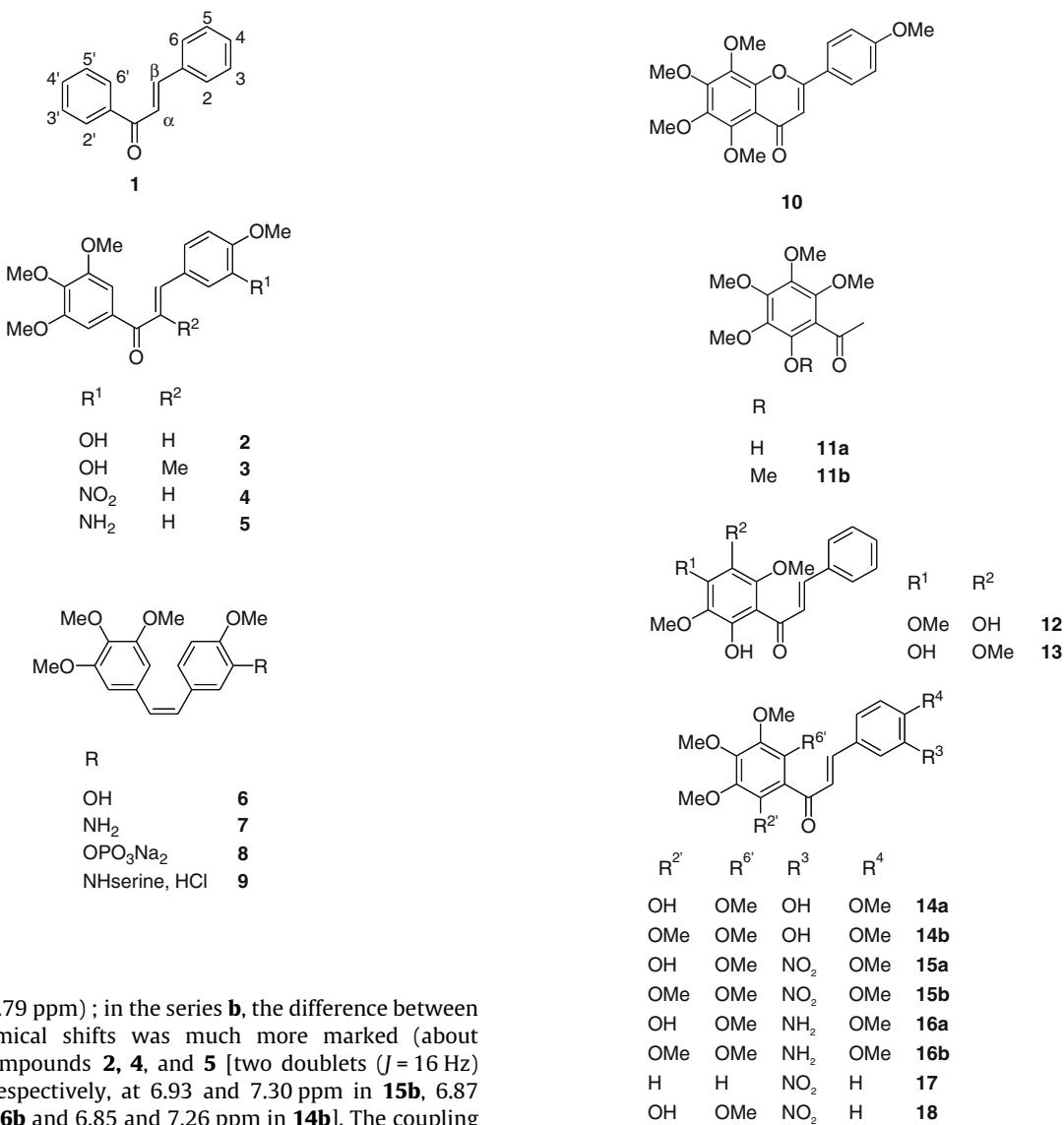
More recently, it was shown that dimethoxylation or trimethoxylation at 2',4',6'-carbons were highly beneficial to cell cycle arrest at G<sub>2</sub>/M, while hydroxylation at 2' was generally detrimental<sup>9</sup> (this last observation was in contrast with previous results showing high cytotoxicity of 2'-hydroxylated chalcones against Jurkat and U 937 cancer cells<sup>10</sup>). So a clear correlation between the polymethoxylation pattern of chalcones on the ring A and their cytotoxicity seems difficult to establish. Starting from tangeretin **10** we used this natural polymethoxylated flavone isolated from *Citrus* for a straightforward access to chalcones peroxygenated on the ring A through the easily available acetophenones **11a** and **11b**. Until now, only two of such chalcones, pedicin **12**, and isodidymo-

carpin **13**, have been studied with regard to cytotoxicity that proved to be weak [(**12**: IC<sub>50</sub> (KB cells) 21.2 μM; IC<sub>50</sub> (Inhibition of tubulin assembly) 300 μM;<sup>11</sup> Compound **13**: IC<sub>50</sub> (P-388 cells) 11.1 μM<sup>12</sup>]. As chalcones **12** and **13** are unsubstituted on the ring B, we undertook synthesis and cytotoxicity evaluation of chalcones **14a** and **14b**, and **16a** and **16b** bearing, respectively, on the ring B the substitution pattern of reference antimitotic chalcones **2** and **5**.

**Chemistry:** Synthesis of these new chalcones was achieved from tangeretin **10** in two, three, or four steps according to scheme 1: (a) basic degradation of **10** into acetophenone **11a**;<sup>13</sup> (b) methylation of **11a** to the pentamethoxyacetophenone **11b**; (c) Claisen–Schmidt condensation of **11a** or **11b** with isovanillin in a mixture MeOH–aqueous KOH giving **14a** and **14b** respectively; (d) same Claisen–Schmidt condensation of **11a** or **11b** but with 3-nitro-4-methoxybenzaldehyde leading to **15a** and **15b**, respectively;<sup>14</sup> (e) reduction of nitrochalcones **15a** or **15b** with SnCl<sub>2</sub> providing the aminochalcones **16a** or **16b**. Reference 3',4',5'-trimethoxylated chalcones **2**, **4**, and **5** were prepared in the same way from 3',4',5'-trimethoxyacetophenone. Lastly, a Claisen–Schmidt reaction between 3-nitrobenzaldehyde and 3',4',5'-trimethoxyacetophenone or acetophenone **11a** led to compounds **17**<sup>13</sup> or **18**, the 4-demethoxylated analogs of 3-nitrochalcones **4** and **15a** (vide infra **Biology**).

From a spectral point of view, a striking difference between series **a** (2'-OH) and **b** (2'-OMe) could be noted in the <sup>1</sup>H NMR spectra: in the series **a**, signals of both olefinic protons were deshielded with chemical shifts for H<sub>α</sub> and H<sub>β</sub> very near [2 doublets (*J* = 16 Hz) at δ 7.75 and 7.85 ppm in **15a**; 2 doublets (*J* = 16 Hz) at δ 7.77 and 7.78 ppm in **16a**], or identical in **14a** (one singlet of

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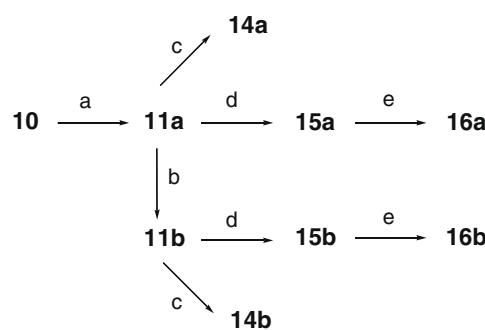


two protons at  $\delta$  7.79 ppm); in the series **b**, the difference between H $\alpha$  and H $\beta$  chemical shifts was much more marked (about 0.4 ppm) as in compounds **2**, **4**, and **5** [two doublets ( $J$  = 16 Hz) for H $\alpha$  and H $\beta$ , respectively, at 6.93 and 7.30 ppm in **15b**, 6.87 and 7.26 ppm in **16b** and 6.85 and 7.26 ppm in **14b**]. The coupling constant of 16 Hz observed in all the spectra (except **14a**) indicates the formation of only the expected *E* isomers. The strong variation of the olefinic signals between series **a** and **b** correlates certainly with a marked difference of conformations, that is related to the presence in the series **a** of an intramolecular hydrogen bonding between the 2'-hydroxyl proton and carbonyl oxygen.

**Biology:** The antiproliferative effect of chalcones was assayed on KB human buccal carcinoma cells and the activation of apoptosis with DEVD-AMC as substrate in HL60 human leukemia cells. Inhibition of tubulin assembly (ITA) was determined according to Zavala and Guenard's method.<sup>15</sup> Compounds were tested at 0.1 mg/ml ( $\approx 2 \times 10^{-4}$  M) and estimated inactive when they decreased by less than 30% the maximum assembly rate of tubulin in the absence of drug. The IC<sub>50</sub> was calculated only for the most active compounds and expressed in relation to deoxypodophyllotoxin (DPPT) in terms of the IC<sub>50</sub>/IC<sub>50</sub> DPPT ratio. As depicted in Table 1, all the chalcones polyoxygenated on the ring A displayed a weak capacity to inhibit cell proliferation (still weaker for compounds **14b**–**16b** of the series **b**). This was in contrast to the strong activity observed with the 3',4',5'-trimethoxylated chalcones **2** and **5**. The measured ITA of the four most active chalcones **2**, **5** and **14a**, **16a** was in good correlation with the cell growth inhibition and matched the activation of apoptosis: **2** and **5** were potent activators (100 nM) compared to **14a** and **16a** (10  $\mu$ M) while **15a** was inactive. Lastly, it was noticed that compounds having the

intramolecular bond between the 2'-hydroxyl and carbonyl function (**14a** and **16a**) exhibit significantly higher ITA than those which do not have this bond (**14b** and **16b**).

In other respects, a recent article reporting some highly selective antileishmanial 3-nitrochalcones led us to evaluate antileish-



**Scheme 1.** Reagents and conditions: (a) EtOH–40% aq KOH, reflux, 5 h, 56%; (b) iodomethane, K<sub>2</sub>CO<sub>3</sub>, DMF, rt, 1 h, quantitative yield; (c) MeOH–50% aq KOH 1:1, isovanillin, rt, 15 h, 31% (**14a**), 64% (**14b**); (d) MeOH–50% aq KOH 10:1, 3-nitro-4-methoxybenzaldehyde, rt, 15 h, 60% (**15a**), 50% (**15b**); (e) SnCl<sub>2</sub>, MeOH, 60 °C, 2 h, 74% (**16a**), 34% (**16b**).

**Table 1**

Antiproliferative, proapoptotic, antitubulin, and antileishmanial activities of synthesized chalcones

Compound	Cytotoxicity on KB cells <sup>a</sup> IC <sub>50</sub> (μM)	Activation of apoptosis in HL60 <sup>b</sup>	ITA activity IC <sub>50</sub> (μM)	Antileishmanial activity on <i>L. donovani</i> promastigotes <sup>f</sup> IC <sub>50</sub> (μM)	Antileishmanial activity on <i>L. amazonensis</i> amastigotes <sup>g</sup> IC <sub>50</sub> (μM)
<b>2</b>	100% IC <sub>50</sub> = 0.017	100 nM: 3.6×	IC <sub>50</sub> = 4.3 1.3 <sup>c</sup>	nd	nd
<b>4</b>	44%		nd <sup>d</sup>	IC <sub>50</sub> = 2.8	Inactive <sup>h</sup>
<b>5</b>	100% IC <sub>50</sub> = 0.031	100 nM: 3.6×	IC <sub>50</sub> = 3.7 1.0 <sup>c</sup>	nd	nd
<b>14a</b>	49%	10 μM: 3.2×	IC <sub>50</sub> = 13 3.8 <sup>c</sup>	IC <sub>50</sub> = 25	Inactive <sup>h</sup>
<b>14b</b>	22%		Inactive <sup>e</sup>	IC <sub>50</sub> = 20	nd
<b>15a</b>	43%	10 μM: 1.2×	Inactive <sup>e</sup>	IC <sub>50</sub> = 5.3	Inactive <sup>h</sup>
<b>15b</b>	19%		Inactive <sup>e</sup>	nd	nd
<b>16a</b>	40%	10 μM: 2.9×	IC <sub>50</sub> = 14 4.1 <sup>c</sup>	IC <sub>50</sub> = 44	Inactive <sup>h</sup>
<b>16b</b>	29%		Inactive <sup>e</sup>	nd	nd
<b>17</b>	29%		nd	IC <sub>50</sub> = 0.4	IC <sub>50</sub> = 26 IS <sup>i</sup> = 2.8
<b>18</b>	31%		nd	IC <sub>50</sub> = 4.4	IC <sub>50</sub> = 26 IS = 2.5

<sup>a</sup> As measured by the MTS assay after 72 h incubation of cells with drug; results are expressed as the percentage of inhibition of cell growth with 10 to 6 M chalcone concentration; IC<sub>50</sub> was calculated only for the two most active compounds.

<sup>b</sup> Activation of caspases 3/7 activity: optimal concentration of compound and fold activation.

<sup>c</sup> IC<sub>50</sub> chalcone/IC<sub>50</sub> deoxypodophyllotoxin.

<sup>d</sup> Not determined.

<sup>e</sup> Estimated inactive when decreasing by less than 30% the maximum assembly rate of tubulin without drug.

<sup>f</sup> As measured by the MTT assay after 72 h incubation of parasite with the drug.

<sup>g</sup> As measured after 30 h incubation of infected macrophages with the drug.

<sup>h</sup> Estimated inactive when amastigotes are still lodged within parasitophore vacuole, inside macrophages, as compared to DMSO control.

<sup>i</sup> Index of selectivity defined by the ratio IC<sub>50</sub> murine macrophages/IC<sub>50</sub> amastigotes.

manial activity of the intermediate nitrochalcones **4** and **15a**.<sup>16</sup> As the same letter displayed also the detrimental influence of almost all substitutions at C-4 on this activity, we decided to compare **4** and **15a** to their 4-unsubstituted analogs **17** and **18**. The antileishmanial activity (Table 1) was determined in vitro against promastigotes of *Leishmania donovani* strain LV9 (MHOM/ET/67/HU3) clone which were grown as described previously,<sup>17</sup> and against intracellular amastigotes of *Leishmania amazonensis* strain LV9 (MPROB/BR/1972/M1841) which were isolated from lesions and purified as described earlier.<sup>18</sup> Cytotoxicity against murine macrophages allowed evaluation of the compound selectivity (Index of selectivity IS = IC<sub>50</sub> macrophages/IC<sub>50</sub> amastigotes). The evaluation on promastigotes of compounds **14a–16a** confirmed the positive effect of a nitro group at C-3 since **15a** is much more active than **14a** and **16a**, but comparison **15a** vs **4** was slightly in favor of **4**. Removing the methoxyl group at C-4 increased the antipromastigotes activity within the 3',4',5'-trimethoxylated series (**17** vs **4**), but not within the series **a** (**18** vs **15a**). In compounds **17** and **18** the lack of substitution at C-4 had a significant impact on efficacy against amastigotes. Only these two compounds lacking substitution at C-4 showed any efficacy in this regard.

In conclusion, our study showed that the peroxygenated substitution patterns of the ring A present in the series **a** and **b** were less beneficial to antiproliferative, activation of apoptosis, antimitotic, and antileishmanial activities than the more classical and available 3',4',5'-trimethoxy substitution. However, this SAR was deduced from only few examples of chalcones, and needs to be generalized to a larger series bearing other substitution patterns on the ring B.

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## References and notes

- Dimmock, J. R.; Elias, D. W.; Beazely, M. A.; Kandepu, N. M. *Curr. Med. Chem.* **1999**, *6*, 1125.
- Go, M. L.; Wu, X.; Liu, X. L. *Curr. Med. Chem.* **2005**, *12*, 483.
- Edwards, M. L.; Stemmerick, D. M.; Sunkara, P. S. *J. Med. Chem.* **1990**, *33*, 1948.
- Ducki, S.; Forrest, R.; Hadfield, J. A.; Kendall, A.; Lawrence, N. J.; McGown, A. T.; Rennison, D. *Bioorg. Med. Chem. Lett.* **1998**, *8*, 1051.
- Lawrence, N. J.; Patterson, R. P.; Ooi, L. L.; Cook, D.; Ducki, S. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 5844.
- Pati, H. N.; Holt, H. L., Jr.; LeBlanc, R.; Dickson, J.; Stewart, M.; Brown, T.; Moses, L. *Med. Chem. Res.* **2005**, *14*, 19.
- LeBlanc, R.; Dickson, J.; Brown, T.; Stewart, M.; Pati, H. N.; VanDerveer, D.; Arman, H.; Harris, J.; Pennington, W.; Holt, H. L., Jr.; Lee, M. *Bioorg. Med. Chem.* **2005**, *13*, 6025.
- Lippert, J. W., III *Bioorg. Med. Chem.* **2007**, *15*, 605.
- Boumendjel, A.; Boccard, J.; Carrupt, P.-A.; Nicolle, E.; Blanc, M.; Geze, A.; Choisnard, L.; Wouessidjewe, D.; Matera, E.-L.; Dumontet, C. *J. Med. Chem.* **2008**, *51*, 2307.
- Rao, Y. K.; Fang, S.-H.; Tzeng, Y.-M. *Bioorg. Med. Chem.* **2004**, *12*, 2679.
- Alias, Y.; Awang, K.; Hadi, H. A.; Thoison, O.; Sévenet, T.; Pa, M. *J. Nat. Prod.* **1995**, *58*, 1160.
- Usman, H.; Hakim, E. H.; Harlim, T.; Jalaluddin, M. N.; Syah, Y. M.; Achmad, S. A.; Takayama, H. Z. *Naturforsch. C* **2006**, *61*, 184.
- Burnham, W. S.; Sidwell, R. W.; Tolman, R. L.; Stout, M. G. *J. Med. Chem.* **1972**, *15*, 1075.
- When the reaction with 3-nitro-4-methoxybenzaldehyde was performed in a mixture EtOH-aqueous KOH (classical conditions), the reaction compounds were a mixture of **15a** and its 4-OEt analog resulting from a SNAr mechanism at C-4.
- Zavala, F.; Guénard, D.; Robin, J.-P.; Brown, E. *J. Med. Chem.* **1980**, *23*, 546.
- Boeck, P.; Bandeira Falcao, C. A.; Leal, P. C.; Yunes, R. A.; Filho, V. C.; Torres-Santos, E. C.; Rossi-Bergmann, B. *Bioorg. Med. Chem.* **2006**, *14*, 1538.
- Desrivot, J.; Waikedre, J.; Cabalion, P.; Herrenknecht, C.; Bories, C.; Hocquemiller, R.; Fournet, A. *J. Ethnopharmacol.* **2007**, *112*, 7.
- Valderrama, J. A.; Zamorano, C.; Florencia Gonzalez, M.; Prina, E.; Fournet, A. *Bioorg. Med. Chem.* **2005**, *13*, 4153.